A METHOD FOR THE STEREOSPECIFIC SYNTHESIS OF THE A AND B RINGS OF GIBBERELLIC ACID E. J. Corey* and Rick L. Danheiser

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A previous report from these Laboratories¹ described the partial synthesis of gibberellic acid (I) from the trienic acid II and recognized III as an attractive precursor to this key intermediate.



This report demonstrates synthetic methodology for effecting the conversion III-II by a stereospecific synthesis of the model system IV. The key step in this sequence is the internal Diels-Alder cyclization of V, conveniently prepared via intermediates VI-XI as outlined below.



Sodium borohydride reduction $(1.2 \text{ equiv}, -24^\circ, 17.5 \text{ hr})$ of 2, 5-diethyl-1-cyclopentanone dicarboxylate VI² gave VII^{3a} in 90% yield. Treatment of VII with 1.1 equiv of methanesulfonyl chloride and 2.5 equiv of 1, 5-diazabicyclo[5.4.0]undec-5-ene (DBU) (benzene, 0.5 hr at 0°, 5 hr at 23°) afforded VIII^{3a} in more than 90% yield⁴ after distillation (bp 107-114°/0.45 mm). Reduction of VIII with aluminum hydride or diisobutylaluminum hydride followed by evaporative distillation from a small amount of anhydrous K_2CO_3 (oven temp 120-130°/0.4 mm) gave IX^{3a} in 50-70% yield.

Manganese dioxide oxidation of IX gave the α , β -unsaturated aldehyde X^{3a} which was transformed to the diene XI^{3a} by treatment with 3 equiv of methylenetriphenylphosphorane in THF (0°, 1.5 hr). Esterification of XI with propiolic acid was accomplished using N,N'-dicyclohexylcarbodiimide in methylene chloride. The ester V³ was obtained in 35% overall yield from VIII in this manner.

The diene ester V underwent intramolecular cycloaddition⁵ above 135° affording the lactone XII^{3a} . Yields of up to 70% were obtained by refluxing V in acetic anhydride⁶ for 20 hr. The lactone XII was best converted to XIII without purification because of its propensity toward aromatization. This was readily accomplished by adding XII to a solution of 1 equiv of lithium isopropylcyclohexylamide⁸ and 3 equiv of hexamethylphosphoramide in THF at -78° and treating the reaction mixture with 1.3 equiv of methyl iodide (-78° to 23° over 7 hr). Isolation and purification by column chromatography on florisil afforded the tricyclic lactone XIII³ mp 60.5-61° in 40-50% overall yield from V. The stereochemistry of XII and XIII follow unambiguously from the method of synthesis.



Oxidation of XIVa could not be achieved without substantial relactonization; XIVb was therefore bromolactonized to permit the required operations at ring B. An aqueous solution of XIVb, obtained by saponification of XIII (KOH, anhydrous CH3OH, 23°, 6.5 hr) and neutralization to pH 7 at 0° with CO_2 , was treated with KBr₃ (1 equiv, -10°, 30 sec) affording a mixture^{3a} of bromolactones XVa and XVb. Oxidation of this mixture with Collins reagent (-23°, 1.5 hr) provided the aldehydes XVIa, b and XVIIa, b, a mixture of epimers. Preparative layer chromatography on silica gel effected purification and concomitant epimerization (ca. 90% as judged by NMR⁹) affording XVIIa, b in 40-56% overall yield from XIII. Oxidation of XVIIa, b with Jones reagent¹⁰ (0°, 30 sec) and esterification of the crude acid with excess CH_2N_2 (ether, 0°) gave the methyl esters XIXa, b^{3a}. Treatment of XIXa, b with excess activated zinc¹¹ (ethanol, 23°, 2 hr) provided the dienic acid IVa³ mp 75-81° in 63% overall yield from XVII.

Alternatively, the acids XVIIIa, b may be protected as p-phenylphenacyl esters for subsequent elaboration of the A ring. This allows for convenient purification and analysis by thin layer chromatography. Reaction of XVIIIa, b with 1 equiv of p-phenylphenacyl bromide and 1 equiv triethylamine (DMF, 0°, 1 hr)^{14a} yielded the esters XXa, b^{3a} which were selectively cleaved to IVb^{3a} upon treatment with excess zinc-silver couple¹⁵ and 1 drop of glacial acetic acid¹² (5:1 ether-ethanol, 23°, 4.5 hr, 55% overall yield from XVIIa, b). Cleavage of phenacyl esters may be readily accomplished in high yield by exposure to 1 equiv of LiSPr in HMPA for 5 min at 0°.¹⁴

An investigation of routes to the key intermediate III is now in progress.¹⁶

References

- 1. E. J. Corey, T. M. Brennan, and R. L. Carney, J. Amer. Chem. Soc., 93, 7316 (1971).
- Obtained in 60% yield by the condensation of diethyl sodiomalonate and diethyl 1, 1-cyclopropanedicarboxylate; see, R. W. Kierstead, R. P. Linstead, and B. C. L. Weedon, J. Chem. Soc., 3616 (1952).
- 3. (a) The proton magnetic resonance, infrared, and (b) high resolution mass spectral data were totally consistent with the assigned structure.
- Previously available by lengthier, less efficient routes: (a) B. L. Nandi, J. Ind. Chem. Soc., <u>11</u>, 277 (1934); (b) H. Prinzbach and H. D. Martin, <u>Chimia</u>, <u>23</u>, 37 (1969); (c) H. C. Stevens, J. K. Rhinehart, J. M. Lavanish, and G. M. Trenta, <u>J. Org. Chem.</u>, <u>36</u>, 2780 (1971); (d) J. D. Roberts, F. O. Johnson, and R. A. Carboni, <u>J. Amer. Chem. Soc.</u>, <u>76</u>, 5692 (1954).
- For other examples of intramolecular Diels-Alder reactions of molecules possessing acetylenic dienophiles see: (a) L. H. Klemm, K. W. Gopinath, G. C. Karaboyas, G. L. Capp, and D. Hsu Lee, <u>Tetrahedron</u>, <u>20</u>, 871 (1964); (b) L. H. Klemm, K. W. Gopinath, D. Hsu Lee, F. W. Kelley, E. Trod, and T. M. McGuire, <u>ibid.</u>, <u>22</u>, 1797 (1966); (c) L. H. Klemm, D. Hsu Lee, K. W. Gopinath and C. E. Klopfenstein, <u>J. Org. Chem.</u>, <u>31</u>, 2376 (1966).
- 6. Several crystals of 4,4'-thiobis-(6-t-butyl-3-methylphenol)⁷ were added as a radical inhibitor.
- Y. Kishi, M. Aratani, H. Tanino, T. Fukuyana, T. Goto, S. Inoue, S. Sugiura, and H. Kakoi, <u>Chem. Commun.</u>, 64 (1972).
- 8. The alkylation of α , β -unsaturated esters has been described: M. W. Rathke and D. Sullivan, <u>Tetrahedron Lett.</u>, 4249 (1972).

- 9. Complete epimerization could be effected by reaction with 0.09 equiv DBU in THF (23°, 17 hr).
- 10. The notoriously acid-sensitive allylic-bridgehead-hydroxyl moiety of the gibberellin C-D rings would require protection at this stage of a total synthesis.
- 11. L. F. Fieser and M. Fieser, "Reagents for Organic Synthesis", John Wiley and Sons, Inc., New York, N. Y., Vol. I p. 1276.
- 12. Neither HCl activated zinc¹¹ nor zinc-copper couple¹³ effected this selective cleavage under a variety of conditions.
- 13. W. T. Brady, H. G. Liddell, W. L. Vaughn, J. Org. Chem., 31, 626 (1966).
- 14. (a) Phenacyl esters have been cleaved using NaSPh at room temperature: J. C. Sheehan, and G. Doyle Davies, <u>J. Org. Chem.</u>, <u>29</u>, 2006 (1964); (b) LiSPr cleaves methyl esters: W. S. Johnson and P. A. Bartlett, <u>Tetrahedron Lett.</u>, 4459 (1970); (c) the conversion of methyl gibberellate to GA₃ with LiSPr has been accomplished.¹
- 15. Prepared by the method of J. M. Conia, J. M. Denis, C. Girard, Synthesis, 549 (1972).
- 16. This work was assisted financially by the National Institutes of Health and the National Science Foundation.